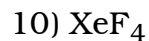
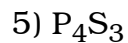
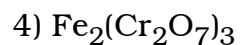
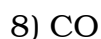
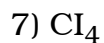


## Naming Molecular Compounds and Ionic Review

Name: \_\_\_\_\_

Write the name of the following compounds:



Write the formula for the following compounds:

1) Lead (IV) Thiosulfate

6) Sulfur Trioxide

2) Chlorine Dioxide

7) Gallium (III) Cyanide

3) Nitrogen Monoxide

8) Dihydrogen Monoxide

4) Carbon Silicate

9) Vanadium (III) Bisulfate

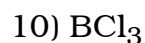
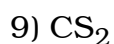
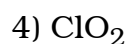
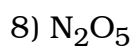
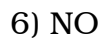
10) Dinitrogen Monoxide

10) Calcium Benzoate

## Naming Molecular Compounds

Name: \_\_\_\_\_

Write the names of the following molecular compounds. Then, tell whether the bonds are polar covalent or nonpolar covalent.



Write the formulas for the following molecular compounds. Then, tell whether the bonds are polar covalent or nonpolar covalent.

1) Tetraphosphorus Decaoxide

6) Dinitrogen Monoxide

2) Trisilicon Octahydride

7) Sulfur Dioxide

3) Dihydrogen Monoxide

8) Sulfur Hexafluoride

4) Carbon Tetraiodide

9) Tetraphosphorous Trisulfide

5) Iodine Pentafluoride

10) Silicon Tetrafluoride