Matter & Density

Name: _____

Part 1: Find the missing unit for the objects using the following information. Show work, and make sure you have the right units:

- 1) mass = 68 g, volume = 29 ml density = _____
- 2) m = 98.354 g, volume = 890.12 ml d = _____
- 3) 454 g, 34 g/cm³ _____
- 4) density = 91 g/cm³, mass = 91 cg
- 5) 1098 milliliters, 568 grams _____
- 6) 9140 ml, 78 g/ml
- 7) mass = 3.46 g, volume = 12.2 mL, density = _____

Part 2: List the state of matter (solid, liquid, gas) and the type of matter [element, compound, solution (homogeneous) colloid or suspension (both heterogeneous)] for each material. Explain why you chose the type of matter you did

material. Explain why you chose the type of matter you did.			
Material	State of Matter	Type of Matter	Reason
Salt Water			
Pure Water			
Jell-O			
Copper			
Wire			
Salt			
Orange			
Juice			
Helium			