## Solubility Graphs

Answer the following questions based on the graph below:

1) What kind of solution would be present if 60 grams of $\mathrm{KNO}_{3}$ were dissolved in 100 g of water at $40^{\circ} \mathrm{C}$ ?
2) If 45 grams of salt were placed in 100 g of water at $80^{\circ} \mathrm{C}$, what type of solution is it?
3) What kind of solution would contain 120 grams of KCl in 300 grams of water at $60^{\circ} \mathrm{C}$ ?

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4) How many grams of NaCl can dissolve in 100 g of water at $20^{\circ} \mathrm{C}$ ?
5) Estimate how much KCl could be dissolved in 300 g of water at $60^{\circ} \mathrm{C}$ to make a saturated solution?
6) Suppose 80 g of KCl were added to 200 g of water at $45^{\circ} \mathrm{C}$. What type of solution would that be?
7) Up to how much NaCl could be dissolved in 25 g of water at $75^{\circ} \mathrm{C}$ ?
8) You have 500 grams of $80^{\circ} \mathrm{C}$ water. How much $\mathrm{KNO}_{3}$ could you dissolve in the solution?

